

Density Functional Theory (DFT) -I

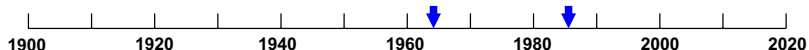
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1

Basis of Most Modern Calculations Density Functional Theory

- Hohenberg-Kohn; Kohn-Sham – 1965
 - A new approach to the interacting electron problem
 - using independent electron methods!
 - Kohn & Sham proposed the LDA
 - the local density approximation - still widely used
- Improved approximations for the density functionals
 - Generalized Gradient Approximations, Orbital dependent functionals, van der Waals functionals, . . .
- Evolution of computer power
- Car-Parrinello Method – 1985
- Nobel Prize for Chemistry, 1998, Walter Kohn



2

Basis of Most Modern Calculations Density Functional Theory

- Why is Density Functional Theory so widely used?
- A new approach to the many-body interacting electron problem
 - In principle DFT is exact
 - Kohn and Sham introduced the idea of an auxiliary system that in principle gives the exact ground state density and energy - but NOT other properties.
 - In practice, DFT allows useful approximations

3

First: What is needed to define the independent particle equation

- From previous lecture:

$$[- (\hbar^2/2m) \nabla^2 + V(\mathbf{r})] \Psi(\mathbf{r}) = E \Psi(\mathbf{r})$$

- where
 - m = mass of particle
 - $V(\mathbf{r})$ = potential energy at point \mathbf{r} = $(d^2/dx^2 + d^2/dy^2 + d^2/dz^2)$
 - E = eigenvalue = energy of quantum state
 - $\Psi(\mathbf{r})$ = wavefunction
 - $n(\mathbf{r}) = |\Psi(\mathbf{r})|^2$ = probability density

- What is needed to specify the equation?

m – fixed for electrons

Boundary conditions

$V(\mathbf{r})$

- Consider m , boundary conditions to be given

4

First: What is needed to define the independent particle equation

- How would you find $V(r)$?

$$[-(\hbar^2/2m)\nabla^2 + V(\underline{r})]\Psi(\underline{r}) = E\Psi(\underline{r})$$

- This is the first problem - which $V(r)$ corresponds to a given solid?
- Some estimate – like $V(r)$ = sum of atomic potentials
- Need a variational principle
- Must be based upon the energy of the system -- the real system including many-body interactions – to be relevant to the real problem

5

Many-Body Electron Problem

The many-electron wavefunction is a function in $3N$ dimensional space

$$\Psi = \Psi(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_N) \quad (2)$$

The total energy is the expectation value

$$E = \frac{\langle \Psi | \hat{H} | \Psi \rangle}{\langle \Psi | \Psi \rangle} \equiv \langle \hat{H} \rangle = \langle \hat{T} \rangle + \langle \hat{V}_{int} \rangle + \int d^3r V_{ext}(\mathbf{r})n(\mathbf{r}). \quad (3)$$

The ground state wavefunction Ψ_0 is the state with lowest energy that obeys the symmetries of the particles and conservation laws.

$$E_0 = \min \frac{\langle \Psi | \hat{H} | \Psi \rangle}{\langle \Psi | \Psi \rangle} \quad (4)$$

electron-electron interactions

potential due to nuclei, other sources "external" to electrons

7

The Fundamental Hamiltonian

Interacting electrons in an external potential

$$\hat{H} = - \sum_i \frac{\hbar^2}{2m_e} \nabla_i^2 - \sum_{i,I} \frac{Z_I e^2}{|\mathbf{r}_i - \mathbf{R}_I|} + \frac{1}{2} \sum_{i \neq j} \frac{e^2}{|\mathbf{r}_i - \mathbf{r}_j|}$$

$$- \sum_I \frac{\hbar^2}{2M_I} \nabla_I^2 + \frac{1}{2} \sum_{I \neq J} \frac{Z_I Z_J e^2}{|\mathbf{R}_I - \mathbf{R}_J|}$$

- Only one small term: The kinetic energy of the nuclei
- If we omit this term, the nuclei are a fixed external potential acting on the electrons
- The final term is essential for charge neutrality – but is a classical term that is added to the electronic part

Not included in following derivations – must be added later

6

The Hohenberg-Kohn Theorem – I

- Hohenberg-Kohn “theorem”(1964)

Schrödinger

$$\begin{array}{ccc} V_{ext}(\mathbf{r}) & \xleftarrow{\text{HK}} & n_0(\mathbf{r}) \\ \downarrow & & \uparrow \\ \Psi_i(\{\mathbf{r}\}) & \Rightarrow & \Psi_0(\{\mathbf{r}\}) \end{array}$$

- All properties of the many-body system are determined by the ground state density $n_0(\mathbf{r})$
- Each property is a functional of the ground state density $n_0(\mathbf{r})$ which is written as $f[n_0]$
- A functional $F[f(\mathbf{r})]$ maps a function to a result: $f(\mathbf{r}) \rightarrow F$

8

The Hohenberg-Kohn Theorem – Ia First Step – definition of functional

- What is a functional? A functional derivative?

Examples

$$I[f] = \int dx \exp(f(x)), \text{ for example, } I[f] = 1/a \text{ for } f(x) = -a^2 x^2$$

$$H[f, g] = \int dx g(x) f(x), \quad \delta H / \delta g(x) = \int dx \delta(x) f(x) = f(x)$$

↖ $\delta = \text{delta function}$

9

The Hohenberg-Kohn Theorem – Ib Legendre Transformation - Example

- The “theorem” is really only a Legendre transformation
- Well-known example: thermodynamic equilibrium
 - Consider a classical liquid – many interacting atoms in thermal motion
 - Energy $E(V)$, $V = \text{volume}$ (where $n = \text{density} = N/V$)
 - What is the equilibrium volume under pressure?

This problem has the classic form of a Legendre transformation:

Define the generalized function:

$$F(P, V) = E(V) + PV + H(P)$$

$$\text{Minimization: } \partial F / \partial V = dE/dV + P = 0 \Rightarrow P \text{ is the pressure } P = -dE/dV$$

$$\partial F / \partial P = dH/dP + V = 0 \Rightarrow V \text{ is the volume } V = -dH/dP$$

Thus the enthalpy $H(P) = E(V) + PV$ where P is the pressure $P = -dE/dV$

This is a Legendre transformation – a change of variables from V to P :

Question for you: Show that H is a function of P and NOT a function of V

10

The Hohenberg-Kohn Theorem – Ic Example of Liquid continued

- So what is the functional for the classical liquid?
- We want to find the state of thermodynamic equilibrium for classical liquid:
 - The equilibrium state of the system at pressure P (an externally applied quantity) can be found by a two-step process
 - Minimize the energy as a function of all internal variables (the many-body problem!) at fixed density $n = N/V \rightarrow$ function $E(n)$,
Note: $E(n)$ is “universal” – it depends only on internal quantities
 - Vary $V \Rightarrow P(V)$ to find the minimum enthalpy H for fixed applied P_{ap}

$$\delta H / \delta P = 0 \Rightarrow H(P) = E(V_0) + P_{ap} V_0 \quad (\text{where } V_0 = N / n_0)$$
 (really very simple interpretation) $P = -dE/dV = P_{ap}$
- Thus $P \rightarrow n_0$ and $n_0 \rightarrow P \rightarrow E$ and all properties!

11

The Hohenberg-Kohn Theorem – II Density functional for electrons

- So what is the functional for electrons? More later on functional
- Consider the Hohenberg-Kohn functional for quantum electrons:

$$E = T[V_{ext}] + V_{int}[V_{ext}] + \int dr n(r) V(r) + F_{HK}[n]$$

$T = KE$, $V_{int} = \text{Expectation value of e-e interactions from Schr. Eq with } V_{ext}$

Hohenberg- Kohn functional

- 1. Minimize the energy as a function of density $n(r)$
- Function of internal variables -- internal variables
- solving the many-body problem! $\rightarrow V = V_{ext}(r)$
- 2. Leads to HK functional $F[n(r)]$
Note: $F[n(r)]$ is “universal” – it depends only on internal quantities
- Thus $V_{ext}(r) \rightarrow n_0(r)$ and $n_0(r) \rightarrow V_{ext}(r) \rightarrow E$ and all properties!

12

A Nobel prize for a Legendre Transformation ???!

- This the first step made by Hohenberg and Kohn sets up the next step – even if it is just a Legendre transformation
- It appears that we have gained nothing! We still have to solve the many body Schrodinger Eq. to get the density functional!
- It is the next step by Kohn & Sham that is truly ingenious :

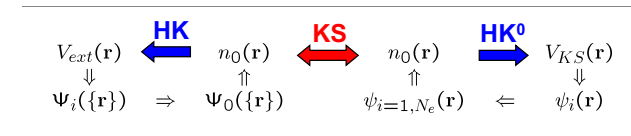
The combination won the Nobel prize

13

The Kohn-Sham Auxiliary System I

- Kohn-Sham (1965) – Replace original many-body problem with an independent electron problem – that can be solved!
- The ground state density is required to be the same as the exact density

$$n_0(\mathbf{r}) = \sum_{\sigma} \sum_{i=1} |\psi_i^{\sigma}(\mathbf{r})|^2,$$



- Only the ground state density and energy are required to be the same as in the original many-body system

15

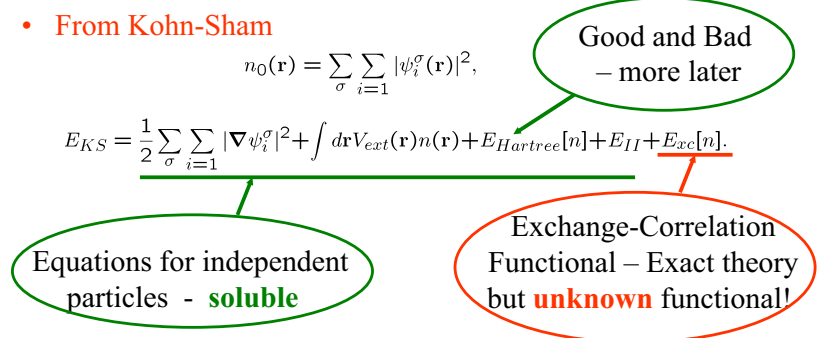
The Hohenberg-Kohn Theorem – III

- The Schrodinger Eq. gives all properties as functional of the potential.
- Why is it a good idea to make a functional of the density instead of a functional of the potential ?
- The hard part of the problem is the correlation energy
 - Correlation means that the probability of finding two electrons at a distance r is reduced for small distances, since the interaction is repulsive
 - This is intrinsically a function of the density and only a secondary effect of the potential
- Thus the first step made by Hohenberg and Kohn sets up the next step – even if it is just a Legendre transformation
- It is the next step by Kohn & Sham that makes DFT practical!

14

The Kohn-Sham Auxiliary System II

- From Hohenberg-Kohn the ground state energy is a functional of the density $E_0[n]$, minimum at $n = n_0$



- The new paradigm for many-body problems
 - Determine only the total energy and density, not other properties
 - find useful, approximate functionals

16

The Kohn-Sham Auxiliary System III

- Kohn-Sham is different from Hohenberg-Kohn

$$E_{KS} = \frac{1}{2} \sum_{\sigma} \sum_{i=1} |\nabla \psi_i^{\sigma}|^2 + \int d\mathbf{r} V_{ext}(\mathbf{r}) n(\mathbf{r}) + E_{Hartree}[n] + E_{II} + E_{xc}[n].$$

- Kohn-Sham introduced the kinetic energy expressed as a functional of wavefunctions of the independent-particle auxiliary system –
- To be precise this is: “Kohn-Sham Density Functional Theory”
- This is one aspect makes KS-DFT practical: The KE is very “non-local” --- Exc is much more local and it can be approximated much more easily the Hohenberg-Kohn functional $F[n]$ which includes the total KE
- The second is that KE, and Coulomb terms are all LARGE These terms must be a good approximation or the method is useless

17

The Kohn-Sham Equations I

- Assuming a form for $E_{xc}[n]$
- Minimizing energy (with constraints) → Kohn-Sham Eqs.

$$n_0(\mathbf{r}) = \sum_{\sigma} \sum_{i=1} |\psi_i^{\sigma}(\mathbf{r})|^2,$$

$$E_{KS} = \frac{1}{2} \sum_{\sigma} \sum_{i=1} |\nabla \psi_i^{\sigma}|^2 + \int d\mathbf{r} V_{ext}(\mathbf{r}) n(\mathbf{r}) + E_{Hartree}[n] + E_{II} + E_{xc}[n].$$

Variational principle

$$\frac{\delta E_{KS}}{\delta \psi_i^{\sigma*}(\mathbf{r})} = 0, \quad (1)$$

$$\langle \psi_i^{\sigma} | \psi_j^{\sigma'} \rangle = \delta_{i,j} \delta_{\sigma,\sigma'} \quad (2)$$

Constraint
Exclusion principle for independent particles

$$\left(-\frac{1}{2}\nabla^2 + V_{KS}^{\sigma}(\mathbf{r}), -\varepsilon_i^{\sigma}\right) \psi_i^{\sigma}(\mathbf{r}) = 0 \quad (3)$$

See Next Slide)

$$\begin{aligned} V_{KS}^{\sigma}(\mathbf{r}) &= V_{ext}(\mathbf{r}) + \frac{\delta E_{Hartree}}{\delta n(\mathbf{r}, \sigma)} + \frac{\delta E_{xc}}{\delta n(\mathbf{r}, \sigma)} \\ &= V_{ext}(\mathbf{r}) + V_{Hartree}(\mathbf{r}) + V_{xc}^{\sigma}(\mathbf{r}) \end{aligned} \quad (4)$$

18

The Kohn-Sham Equations II

- Kohn-Sham Self-Consistent Equations.
- Assuming a form for $E_{xc}[n]$

$$\left(-\frac{1}{2}\nabla^2 + V_{KS}^{\sigma}(\mathbf{r}), -\varepsilon_i^{\sigma}\right) \psi_i^{\sigma}(\mathbf{r}) = 0 \quad (3)$$

$$\begin{aligned} V_{KS}^{\sigma}(\mathbf{r}) &= V_{ext}(\mathbf{r}) + \frac{\delta E_{Hartree}}{\delta n(\mathbf{r}, \sigma)} + \frac{\delta E_{xc}}{\delta n(\mathbf{r}, \sigma)} \\ &= V_{ext}(\mathbf{r}) + V_{Hartree}(\mathbf{r}) + V_{xc}^{\sigma}(\mathbf{r}) \end{aligned} \quad (4)$$

Kohn-Sham Potential
Function position \mathbf{r}
Functional of density

Result
Density, Energy

$$n_0(\mathbf{r}) = \sum_{\sigma} \sum_{i=1} |\psi_i^{\sigma}(\mathbf{r})|^2,$$

$$E_{KS} = \frac{1}{2} \sum_{\sigma} \sum_{i=1} |\nabla \psi_i^{\sigma}|^2 + \int d\mathbf{r} V_{ext}(\mathbf{r}) n(\mathbf{r}) + E_{Hartree}[n] + E_{II} + E_{xc}[n].$$

Eigenvalues are approximation to the energies to add or subtract electrons –electron bands
More later

Uses Chain rule
 $dE/d\psi = (dE/dn) (dn/d\psi)$

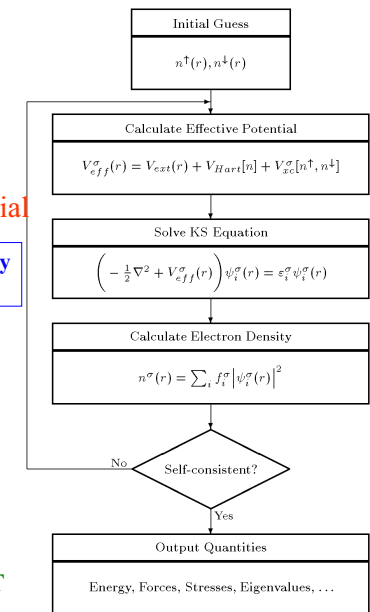
19

Solving Kohn-Sham Equations

- Structure, types of atoms
- Guess for input density or potential
- Solve KS Eqs. This is the numerically intensive part
- New Density and Potential
- Self-consistent?
- Output:
 - Total energy, force, stress, ...

KS Eigenvalues - Only parameters - NOT eigenvalues of original man-body system

Self-Consistent Kohn-Sham Equations



20

Why are forces and stresses given correctly ?

- The force on a nucleus $\mathbf{F} = -d\mathbf{E}/d\mathbf{R}$ is given by the “force theorem” also called the “Helmann-Feynman” theorem
 - For a small perturbation $d\mathbf{R}$, you might think this requires expansion in the excited` states
 - **Exercise for you – Prove that \mathbf{F} is given strictly by the density including all many-body terms**
- Stress $\sigma = -dE/d\varepsilon$ (generalization of virial theorem)
 - The same reasoning
- Perhaps most surprising – second derivatives (and higher) are also exact in principle.
 - Extension of above reasoning
 - Different from textbook perturbation theory – **Why?**