Electronic Structure of Condensed Matter Fudan University - April 2010

Lecture 1 Introduction and Overview

Instructor: Richard M. Martin University of Illinois Stanford University

RMartin@illinois.edu

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Outline for Course

- Qualitative concepts electrons in materials
 - Molecules, solids, liquids, . . .
 - Here we emphasize solids Examples, unique features of solids
- Basic theory and methods
 - Calculations for real materials possible only because of the power of computers and development of algorithms
 - But much more important is the fundamental developments that are the basis for all present-day (and future!) methods
- Calculations of the properties of materials
 - Theoretical relations that make possible quantitative calculations
 - Working hand-in-hand with experimentalists
 - Important: The most used methods do not always work! Crucial to know how to recognize errors and inaccuracies!
- Developments of many-body methods
 - Topics of special lectures

Contents of a typical solid state physics text (Kittel, Ashcroft and Mermin,)

Structures of crystals

1

2

- Definitions of structures
- Typical bonding and structures
- Structural phase transitions
- Mechanical Properties
 - Lattice vibrations phonons
 - Elastic constants
 - Piezoelectric constants
 - Effective charges
- Thermal Properties
 - Heat capacity, heat conduction
- Electronic properties
 - Definition of bands
 - Metals vs. insulators
 - Conductivity, dielectric functions

Properties that can be described without knowing anything about electrons

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So why are they a part of electronic structure?

Properties that can only be understood in terms of the electrons?

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Aspects of course

- Material on Web site: schedule, links to files
 - PDF files for lectures posted after the lecture (Thursday afternoon or Friday this week)
- Three sets of homework exercises
- Project of your choice but approved by me in advance
 - time required approximately equal to 2 sets of exercises)

Electronic Structure Theory and Computation Properties of Materials

All properties of materials are controlled by the electrons interacting with the nuclei

The effects can be divided into two categories

- Electrons form the bonds that hold the nuclei together
 - The structures of molecules and solids
 - Strength of materials, elastic constants, vibration frequencies, . . .

Thermal properties, melting, liquids, . . .
 Determined by the ground state of the electrons

Electrons determine

– Example: Superconductivity

- Many other examples

- Electrical conductivity: Insulators, Metals, Semiconductors, Superconductors
- Optical properties, dielectric constant, colors, ...

Magnetism - a purely quantum effect – due to the electrons
 Determined by the excited states of the electrons

Electronic Structure Theory and Computation

Properties of Materials

If all properties of materials are determined by the electrons.

- Theoretical and experimental methods that are fields in themselves

Electronic Structure has an important role

the first questions were:

- Calculation of electron-phonon interactions

What is not "Electronic structure"?

Recent developments actually predict transition temperatures

- In all the recent discoveries (High-Tc, C₆₀, MgB₂, Fe pnictides)

- Nevertheless, the phenomena are an entire field of research

What is the electronic structure? What does "LDA" predict?

Electronic Structure in Perspective A brief History

A long way in less than 90 years



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L. de Broglie – Nature 112, 540 (1923).



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- Pauli exclusion Principle 1925
- Fermi statistics 1926
 - Independent Electron Approximation
- Thomas-Fermi approximation 1927 (First density functional Dirac 1928
- Dirac equation relativistic quantum mechanics 1928
- Bloch theorem 1928
- Wilson Implications of band theory Insulators/metals -1931
- Wigner- Seitz Quantitative calculation for Na 1935
- Slater Bands of Na 1934 (proposal of APW in 1937)
- Bardeen Fermi surface of a metal 1935

Not included here

1900	1920	1940 Transisto	1960 or	1980	2000	2020 8

A long way in less than 90 years

The basic methods of electronic structure

- Slater Augmented Plane Waves (APW) 1937
 Not used in practice until 1950's, 1960's electronic computers
- Herring Orthogonalized Plane Waves (OPW) 1940
 First realistic bands of a semiconductor Ge Herrman, Callaway (1953)
- Hellman, Fermi Pseudopotentials 1930's
 - Phillips, Kleinman, Antoncik, 1950's
 - Hamann, Vanderbilt, ... 1980's
 - Andersen Linearized Muffin Tin Orbitals (LMTO) 1975
 - The full potential "L" methods LMTO, LAPW

Many-body methods to treat electron-electron interactions

- Recognized since the early days of quantum mechanics (Hylleras $H_2 1929$) The basic advances in many-body theory -1950's - 60's
- Landau Feyman Bohm Pines Gell-Mann Breuckner Hubbard
- Baym Kadanoff

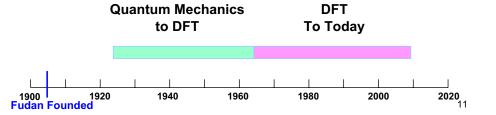


Emphasis in this course

Basic Theory and Practical Methods	Special Lectures (very brief)			
1964-5 Density Functional Theory	Quantum Monte Carlo and "GW"			
★ Hohenberg, Kohn, Sham	McMillan			
Exact ground state energy	QMC - Exact ground state for			
a functional of electron density	interacting Bosons			
Local Density Approx. (LDA)	Hedin " GW ""			
1970's Computation established as p	owerful tools			
Methods using DFT	Fermion QMC (Ceperley,)			
Computational power for precise	Computational power to treat			
calculations in diverse crystals	homogeneous electron gas			
1980's New Methods and New Discoveries	s (STM, QHE, C ₆₀ , Hi-Tc, …)			
★ Car-Parrinello molecular dynamics	QMC Calculations on solid H			
in electronic calculations	Quantitative "GW" for Excited States			
1990's Discoveries and Methods continue (N	lanotubes, CMR, MgB ₂ , …)			
✤ Polarization "Order N"	Dynamical Mean Field Theory			
"Order N"				
2000's Computation – on PC and Superc	computer			
1900 1920 1940 1960	1980 2000 2020 10			

A long way in less than 90 years

Electronic Structure in Perspective A brief History



The Fundamental Hamiltonian

$$\hat{H} = -\sum_{i} \frac{\hbar^{2}}{2m_{e}} \boldsymbol{\nabla}_{i}^{2} - \sum_{i,I} \frac{Z_{I}e^{2}}{|\mathbf{r}_{i} - \mathbf{R}_{I}|} + \frac{1}{2} \sum_{i \neq j} \frac{e^{2}}{|\mathbf{r}_{i} - \mathbf{r}_{j}|}$$
$$-\sum_{I} \frac{\hbar^{2}}{2M_{I}} \boldsymbol{\nabla}_{I}^{2} + \frac{1}{2} \sum_{I \neq J} \frac{Z_{I}Z_{J}e^{2}}{|\mathbf{R}_{I} - \mathbf{R}_{J}|}$$

- Only one small term, the kinetic energy of the nuclei
 - Very good approximation to neglect in determining the electronic states Born-Oppenheimer approximation
- All other terms are large and of the same order of magnitude
 - The difficult part is the electron-electron interaction
 - Cannot be neglected in any quantitative calculation
- Note For simplicity, we do not consider magnetic fields, and we neglect spin orbit and other relativistic

Two types of Goals for Electronic Structure

Understanding

- Qualitative understanding of electronic properties does NOT require large calculations
- Understanding also means understanding the behavior of specific materials that are ultimately determined by the electrons
- Quantitative theoretical and computational methods for properties of materials
 - Truly predictive methods must start from the fundamental equations
 - This requires:

1. Ingenious theoretical methods to treat the interactiing electron problem for large classes of materials

2. Ingenious computational methods to make calculations feasible for real materials

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Lecture 1 - continued Examples related to Chapter 2

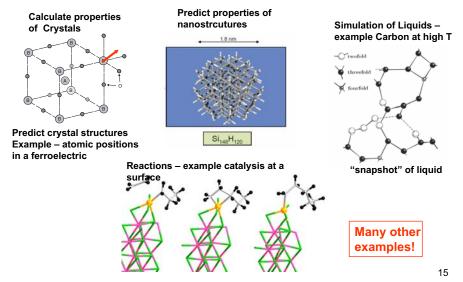
Physical distinction between different materials Delocalized vs, Localized electronic states Narrow vs. Wide bands

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What does one want to do?

Treat many types of materials in a unified way from the basic equations



Examples of Modern Calculations

- Properties of crystals many calculations are now "routine"
 - Definitive tests of the theory comparisons with experiments
- Calculations for complex systems
 - Theory provides key role along with experiments
 - Understanding
 - Predictions
 - Direct simulation of atomic scale quantum phenomena
- Examples

1900

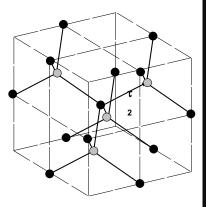
- Surfaces, interfaces, defects,
- Thermodynamic phase transitions, Liquids, Melting, ...
- Nanostructures in real environments, ...
- Large complex molecules in solution,

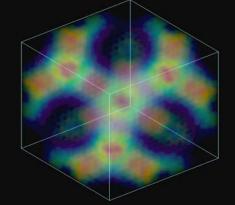
192	20	194	10	190	60	1980



2000

Examples of Modern Calculations Electron density in silicon

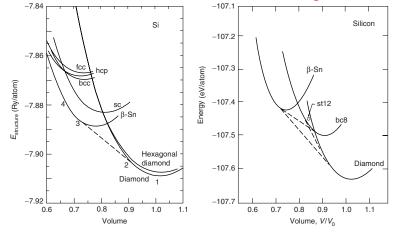




In Si the black and grey atoms are identical Measure of methods – density agrees well with experiments

Phase Transitions Under Pressure

Transformation of silicon to metallic structures under pressure Predictions from LDA – first demonstrations that DFT gives accurate results



Modern codes ((ABINIT, VASP, ESPRESSO, Wein LAPW, FPLO, Beijing code, . . .) can easily reproduce similar curves. Possible project: calculations with various functionals.

Results agree with experiment

(in important classes of materials)

Different methods agree (when done carefully)

Method	С		Si		CaF ₂		bcc Fe		
	a	B	a	B	a	В	a	В	m
$NCPP^a$	3.54	460	5.39	98	5.21	90	2.75 ^c	226 ^{<i>c</i>}	
PAW^a	3.54	460	5.38	98	5.34	100			
PAW^b	3.54	460	5.40	95	5.34	101	2.75	247	2.00
$USPP^b$	3.54	461	5.40	95	5.34	101	2.72	237	2.08
$LAPW^a$	3.54	470	5.41	98	5.33	110	2.72^{d}	245^d	2.04^d
EXP^a	3.56	443	5.43	99	5.45	85-90	2.87^{d}	172^d	2.12^d

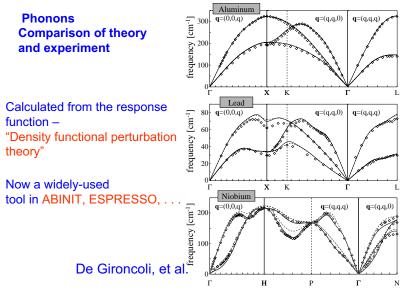
a – lattice constant, B – bulk modulus, m – magnetization

^aHolzwarth, et al.; ^bKresse & Joubert; ^cCho & Scheffler; ^dStizrude, et al. •

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Vibration frequencies – Phonons

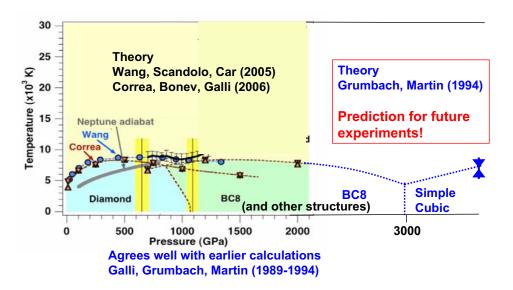


theory"

The Car-Parrinello Advance

- Car-Parrinello Method 1985
 - Simultaneous solution of Kohn-Sham equations for electrons and Newton's equations for nuclei
- A revolution in the power of the methods
 - Relaxation of positions of nuclei to find structures
 - "First principles" molecular dynamics simulations of solids and liquids with nuclei moving thermally
 - Chemical reactions, . . .
 - Stimulated further developments . . .

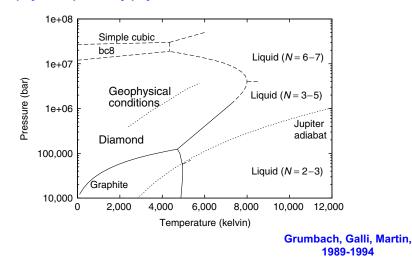
Predicted phase diagram of Carbon Prediction <u>Before</u> Experiment



Unified MD – nuclei and electronic states Example of Carbon

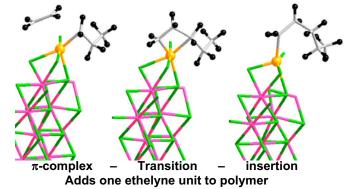
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Melting of diamond at high pressure – other phases - geophysical, planetary physics



Simulation of Catalysis for polymerization

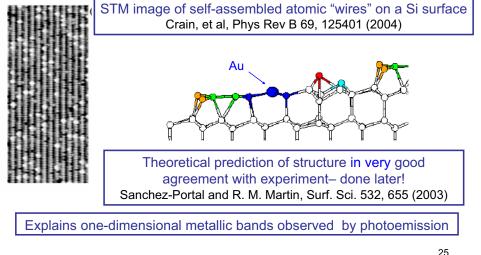
- Unraveling the steps in the Ziegler-Nata reaction
 - Industrial process for production of polyethylene
 - Simulations with Car-Parrinello molecular dynamics M. Boero, et al.



Surfaces – Au on Silicon surfaces

("vicinal" surface at angle to 111)

Atomic scale Au wires on Si (557) surface



So what is left to do?

- Failures of present density functional approximations in important cases!
- Electronic excitations - The "band gap problem"

- . . .

. . .

- Strongly interacting systems
 - Magnetic insulators, metal-insulator transitions, - . . .
- Example of "Failure" that has been solved!
 - Weak Van der Waals bonds
 - Development of non-local functional has the right distance dependence and is accurate enough for real materials

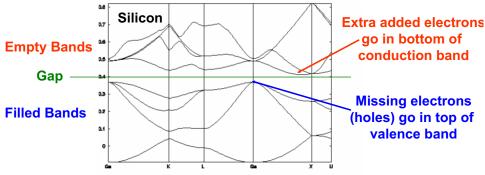
Simulations of DNA (Eårly čalculation – not very accurate) Machado de de la contractione de Full calculations with atoms moved with molecular dynamics A-structure poly dC- poly dG valence ch of

Iso-density surfaces Universitat Autonoma Barcelona -- February 17, 2006

Electron Excitations - Bands

 Understood since the 1920's - independent electron theories predict that electrons form bands of allowed eigenvalues, with forbidden gaps

Established by experimentally for states near the Fermi energy



Electron Excitations – Bands

(Description using Many-Body Methods)

- Excitations
- Electron removal (addition)
 - Experiment Photoemission
- - Theory Quasiparticles"GW" Approximation
 - Discussion in Special topics lectures.
 Theoretical concepts and basic methods, but not details mathematics or methods.

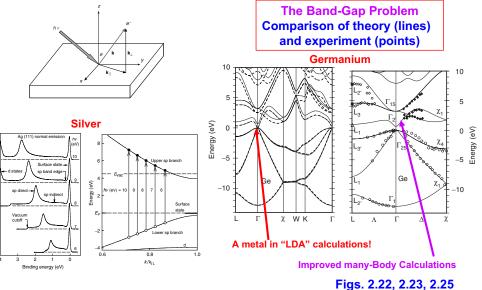
Electron Excitations – Bands

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Experiment

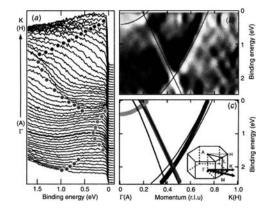
Angle Resolved Photoemission (Inverse Photoemission) Reveals Electronic Removal (Addition) Spectra



Experiment

Angle Resolved Photoemission (Inverse Photoemission) Reveals Electronic Removal (Addition) Spectra

Recent ARPES experiment on the superconductor MgB₂ Intensity plots show bands very close to those calculated



Extended band-like vs. Localized atomic-like behavior of electrons in solids

Very delocalized states

Very localized states

Nearly free electron bands Alkali metals: Na, K, ... 4f states Rare earths: Pr – Th

Intermediate cases

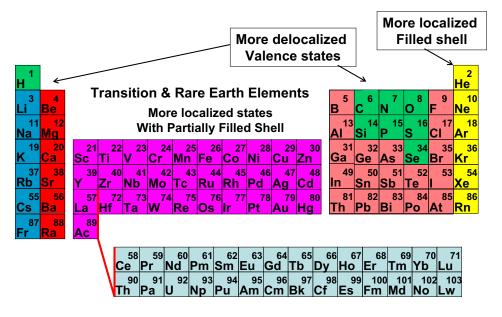
Semiconductors "Band Gap Problem" Transition Metals Magnetism, Met-Ins Trans.

NOT included in text

Mott Metal-Insulator Transitions Hydrogen under pressure (not reached experimentally) NiO, CuO (actual Mott transitions) "Colossal Magneto-Resistance" CMR

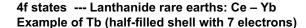
Anderson/Kondo/Heavy Fermion Problems "Anomalous" Rare Earths, e.g., Ce Transition Metal – Semiconductor alloys – Magnetic semiconductors Hi-Tc,

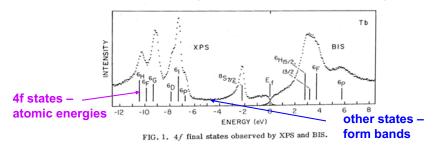
Periodic Table



Examples of Strong Correlations

Review Articles: Imada, et al., Rev. Mod. Phys. 70, 1039 (1998) Damascelli, et al., Rev. Mod. Phys. 75, 473 (2003)





For the 4f states, the energies to add or remove electrons are essentially the same as in an isolated atom – strong interactions on the atom lead to "multiplets" – different ways the same number of electrons on an atom can be arranged

Examples of Strong Correlations

Review Articles: Imada, et al., Rev. Mod. Phys. 70, 1039 (1998) Damascelli, et al., Rev. Mod. Phys. 75, 473 (2003)

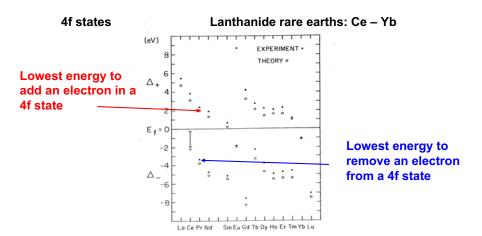
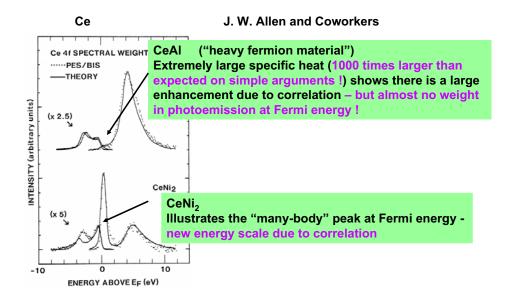


FIG. 2. Comparison of experimental and theoretical (Refs. 11 and 12) values of \triangle_{-} and \triangle_{+} . The experimental \triangle_{-} value for Ce lies within the error bar.

Example of Extreme Enhancements - Ce



Goal of Course

Understanding of the role of electrons in condensed matter

• Practical, useful knowledge of methods that are working tools of theorists, experimentalists, and researchers in many fields Without the burden of heavy math, many-body theory,

• Appreciation for the real many-body problems presented by electrons in condensed matter

Understanding when to expect correlation to be important The grand challenges of condensed matter physics today

For those who are interested

• Practical, useful knowledge of many-body methods that are becoming more and more important for quantitative predictions and understanding the grand challenges in condensed matter